Subject Code	Course Title	L	T	P	C	Total Hours
	ENGINEERING CHEMISTRY	3	1	0	4	45

Unit-I: CHEMICAL THERMODYNAMICS

Entropy - entropy changes in isothermal expansion of an ideal gas - reversible and irreversible processes - work & free energy functions - Helmholtz and Gibbs free energy functions - Gibbs-Helmholtz equation - Gibbs-Duhem equation - Clausius-Clapeyron equation & its applications - Van't Hoff isotherm and its applications.

Unit-II: CHEMICAL KINETICS AND CATALYSIS

Kinetics of second and third order reactions –half life period – saponification of ester – kinetics of opposing, parallel and consecutive reactions and its examples - effect of temperature on reaction rate – theory of absolute reaction rate. Classification and characteristics of catalysts – autocatalysis – steady state principle - enzyme catalysis – Michaelis menton equation (derivation) – acid base catalysis (derivation).

Unit-III: THERMAL AND SPECTROSCOPIC TECHNIQUES

Thermogravimetry (TGA) – schematic and block diagram – characteristics of thermo-balance design – methods expressing TG results – applications in qualitative analysis, composition of alloys and mixtures, study of polymers. Differential thermal analysis (DTA) - schematic and block diagram – representation of DTA data – qualitative application (calcium oxalate monohydrate only). Electromagnetic spectrum – Beer Lambert's law (Derivation) – principle, theory, instrumentation and simple applications of: Flame photometry – UV-visible spectroscopy - IR spectroscopy.

Unit IV: CORROSION - THEORY & PROTECTION

Electrochemical cells – standard electrode potential - electrochemical series - principles of chemical and electrochemical corrosion – factors influencing corrosion – types of corrosion - galvanic corrosion - differential aeration corrosion - stress corrosion – corrosion control - cathodic protection and sacrificial anode – corrosion inhibitors - protective coatings - constituents, functions and uses of paints and varnishes.

Unit-V: POLYMERS AND NANOMATERIALS

Polymer Chemistry: Monomers – functionality – polymers - degree of polymerization – effect of polymer structure on properties – addition, condensation, co-polymerization - mechanism of addition polymerization (free radical polymerization only). Nanomaterials: Introduction - synthesis of nano materials by physical and chemical methods - ball milling - chemical vapour deposition -sol-gel method - applications of nano materials.

Text Book

1. Engineering Chemistry, P.C. Jain and Monika Jain, Dhanpat Rai Publishing Co Pvt. Ltd., New Delhi, 2008. Reference Books

- 1. Principles of Physical Chemistry, B.R. Puri, L.R. Sharma and Madan S. Pathania, Shoban Lal Nagin Chand & Co., Jalandhar, 2000.
- 2. Physical Chemistry for Engineers, P.C. Jain and Renuka Jain, Dhanpat Rai & sons, New Delhi, 2001.
- 3. Applied Chemistry, K. Sivakumar, Anuradha Publications, Chennai, 2009.
- 4. Chemistry in Engineering & Technology, J.C. Kuriacose and J. Rajaram, Vol. 1, Tata McGraw-Hill, New Delhi, 1996.